Electrostatics Review

All matter is made up of atoms, and all atoms are made up of three subatomic particles: <u>Electrons</u>, neutrons, and <u>protons</u>

Draw a diagram of an atom to show the subatomic particles <u>Electrons</u> are negative <u>Protons</u> are positive Neutrons are neutral

When atoms gain electrons they become <u>negative</u> ions, when they lose electrons they become positive ions

Electrostatics is the study of static electricity

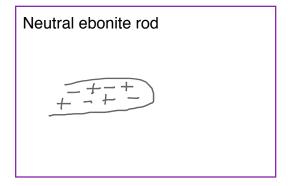
When a static charge is created by rubbing two objects together <u>electrons</u> are not created, rather they are <u>transferred</u> from one object to another. The object with the stronger ability to attract electrons becomes <u>negative</u> and the object with the weaker ability to attract electrons becomes <u>positive</u>. To help predict the charge that will be created by rubbing two objects together we can use the <u>electrostatic</u> series.

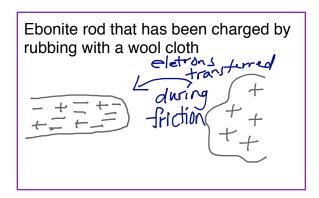
State the law of electric charges: Like charges repel Unlike charges attract

If we have charged objects and we want to determine if they are positively or negatively charged we can use the attraction test. With the attraction test there are **two** ways to see an attraction: <u>between opposite</u> (unlike) charges or a charged object and a neutral object

But there is only one way that there will be repulsion: <u>must be the **same** (like) charge</u> When and object is neutral, it has an <u>equal</u> number of positive and negative charges.

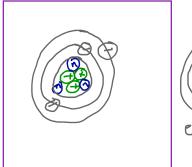
When an object has been charged, electrons have been transferred, so now there is an <u>unequal</u> number of positive And negative charges.





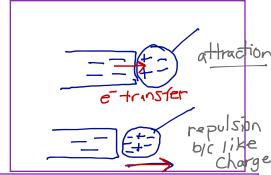
When. Balloon is charged by friction against your hair it becomes <u>negatively</u> charged. Because the balloon is made out of a material that is an <u>insulator</u> the charge stays in one place. A charge that does not move is called <u>static</u> charge. This is an example of charging by <u>friction</u>.

Charges can also be transferred by contact. When a charge is transferred by contact, the charge transferred will be the <u>same</u>.



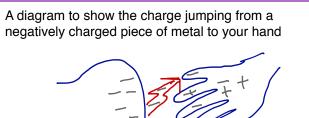


A diagram that shows the negative charge on a polyethylene strip being transferred to a pith-ball electroscope.

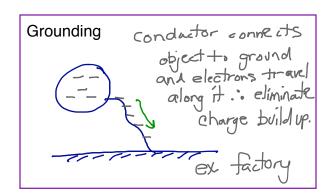


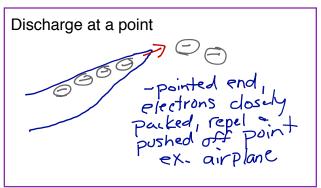
Sometimes a charge can "jump" from one object to another. When this happens we often see a <u>spark</u>. The charge always jumps from an area with a <u>high</u> amount of negative charge to an area with a <u>low</u> amount of negative charge. This can be dangerous because sometimes there is heat created as e <u>electrons</u> move through the air. One potential danger is that when at a gas pump: could ignite gas and cause a fire or <u>explosion</u>.

Example 2: lightning!

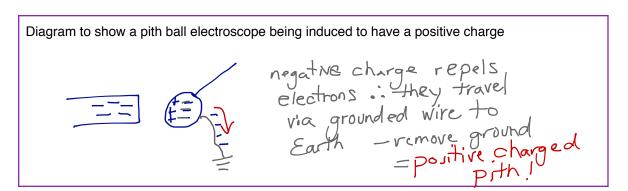


For safety reasons we often want to remove built-up negative charges. The process of removing the negative charges to return the object to neutral is called <u>discharging</u>. To remove the build-up of negative charges there are two main methods, explain how each works:





An electrical charge can also be created without two objects coming into contact. This is known as charging by <u>induction</u>. When a charge is created this way it is the <u>opposite</u> of that on the object used to create the charge. If there is a ground attached to the object being charged, then a <u>permanent charge</u> can be created.



Electric charges can also be built up on conductors. A conductor is a material that allows <u>electrons to flow/ move freely.</u> This means that on a charged conductor the charge is <u>evenly</u> distributed over the surface of the object.

